

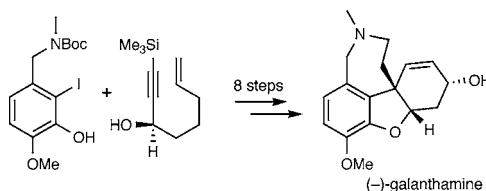
Stereocontrolled Synthesis of
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ABSTRACT



An enantioselective synthesis of (–)-galanthamine has been realized in 11 linear steps starting from isovanillin. A Mitsunobu aryl ether forming reaction was used to assemble the galanthamine backbone, which was stitched together using enyne ring-closing metathesis, Heck, and N-alkylation reactions affording the tetracyclic ring system. Control of relative and absolute stereochemistry was derived from an easily accessible enantiomerically enriched propargylic alcohol 13.

Galanthamine (**1**) is an *Amaryllidaceae* alkaloid,^{1,2} which is in clinical use for the symptomatic treatment of Alzheimer's disease.³ (–)-Galanthamine displays competitive reversible inhibition of acetylcholine esterase (AChE) and allosteric potentiation of nicotinic acetylcholine receptors.^{4,5} These effects are believed to be associated with the therapeutic effect of the drug. (–)-Galanthamine is relatively expensive to obtain from natural sources, leading to the development of a synthetic process for its commercial production. This

process utilizes oxidative phenolic coupling and crystallization-induced chiral conversion as key steps^{6–8} and is effective for the synthesis of the natural product but somewhat limited in terms of analogue synthesis due to the requirement of electron-rich aryl groups. Alternative synthetic routes to galanthamine and structurally related alkaloids have been reported,^{1,3,9–11} including an asymmetric total synthesis.¹²

Recognition of the benefits of a synthetic route to galanthamine with the potential to create structural analogues not available through the oxidative coupling approach^{3,12,13} stimulated our initial synthetic efforts toward the natural product. Analysis of (–)-galanthamine (**1**) led us to consider a diene of general structure **2** as a key intermediate, which contains suitable functionality for the closure of the two

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(1) (a) Hoshino, O. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: New York, 1998; Vol. 51, pp 323–424. (b) Martin, S. F. In *The Alkaloids*; Brossi, A., Ed.; Academic Press: New York, 1987; Vol. 30, pp 251–376. (c) Jin, Z. *Nat. Prod. Rep.* **2003**, *20*, 606–614.

(2) Proskurnina, N. F.; Yakovleva, L. P. *Zh. Obshch. Khim.* **1952**, *22*, 1899–1900.

(3) For a recent review of the synthesis and pharmacology of galanthamine: Marco-Contelles, J.; Carreiras, M. D.; Rodriguez, C.; Villarroja, M.; Garcia, A. G. *Chem. Rev.* **2006**, *106*, 116–133.

(4) Nordberg, A.; Svensson, A.-L. *Drug Safety* **1998**, *19*, 465–480.

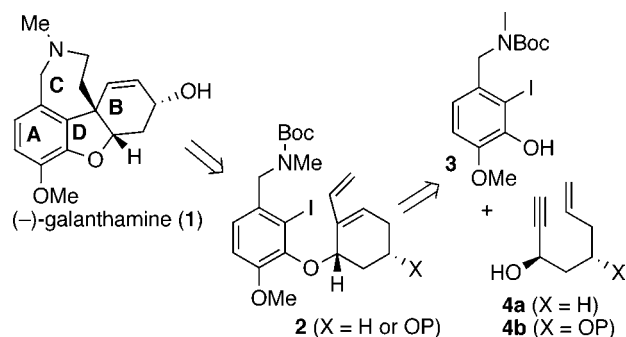
(5) (a) Lilienfield, S. *CNS Drug Rev.* **2002**, *8*, 159–176. (b) Popa, R. V.; Pereira, E. F. R.; Lopes, C.; Maelicke, A.; Albuquerque, E. X. *J. Mol. Neurosci.* **2006**, *30*, 227–232. (c) Maelicke, A.; Albuquerque, E. X. *Eur. J. Pharmacol.* **2000**, *393*, 165–170.

(6) For detailed discussion of the synthesis of galanthamine by oxidative phenolic coupling, see refs 1b and 3. For the first example of this approach, see: (a) Barton, D. H. R.; Kirby, G. W. *J. Chem. Soc.* **1962**, 806–817. For stereoselective oxidative coupling approaches, see: (b) Shimizu, K.; Tomioka, K.; Yamada, S. I.; Koga, K. *Chem. Pharm. Bull.* **1978**, *26*, 3765–3771. (c) Kodama, S.; Hamashima, Y.; Nishide, K.; Node, M. *Angew. Chem., Int. Ed.* **2004**, *43*, 2659–2661.

(7) Küenburg, B.; Czollner, L.; Fröhlich, J.; Jordis, U. *Org. Process Res. Dev.* **1999**, *3*, 425–431.

(8) Total spontaneous resolution of (–)- or (+)-narwedine: Shieh, W.-C.; Carlson, J. A. *J. Org. Chem.* **1994**, *59*, 5463–5465.

Scheme 1. Proposed Synthetic Route to (–)-Galanthamine



heterocyclic C and D rings (Scheme 1). Formation of the D ring, and corresponding quaternary stereogenic center, would be achieved by application of the Heck strategy introduced by Fels and Parsons.^{11,12} The proposal centered on the use of an enyne ring-closing metathesis (RCM) reaction to create the vinyl-substituted cyclohexene ring within a structure containing all of the carbon atoms required in the final target.¹⁴ Joining the two fragments would be effected through Mitsunobu coupling of a phenol **3** with an enantiomerically enriched propargylic alcohol **4a**,¹⁵ readily available through asymmetric ketone reduction. Ultimately, we imagined that such an approach could be adapted to incorporate the galanthamine allylic hydroxyl group within the fragment **4b**.¹⁶

(9) Lycoramine (racemic): (a) Hazama, N.; Irie, H.; Mizutani, T.; Shingu, T.; Takada, M.; Uyeo, S.; Yoshitake, A. *J. Chem. Soc. (C)* **1968**, 2947–2953. (b) Misaka, Y.; Mizutani, T.; Sekido, M.; Uyeo, S. *J. Chem. Soc. (C)* **1968**, 2954–2959. (c) Schultz, A. G.; Yee, Y. K.; Berger, M. H. *J. Am. Chem. Soc.* **1977**, 99, 8065–8067. (d) Martin, S. F.; Garrison, P. J. *J. Org. Chem.* **1981**, 46, 3567–3568. (e) Martin, S. F.; Garrison, P. J. *J. Org. Chem.* **1982**, 47, 1513–1518. (f) Sánchez, I. H.; Soria, J. J.; López, F. J.; Larraza, M. I.; Flores, H. J. *J. Org. Chem.* **1984**, 49, 157–163. (g) Ackland, D. J.; Pinhey, J. T. *J. Chem. Soc., Perkin Trans. I* **1987**, 2695–2700. (h) Parker, K. A.; Kim, H. J. *J. Org. Chem.* **1992**, 57, 752–755. (i) Ishizaki, M.; Ozaki, K.; Kanematsu, A.; Isoda, T.; Hoshino, O. *J. Org. Chem.* **1993**, 58, 3877–3885. (j) Essamkaoui, M.; Benharref, A.; Moskowit, H.; Mayrargue, J.; Thal, C. *Heterocycl. Commun.* **1996**, 2, 319–323. (k) Gras, E.; Guillou, C.; Thal, C. *Tetrahedron Lett.* **1999**, 40, 9243–9244. (l) Fan, C. A.; Tu, Y. Q.; Song, Z. L.; Zhang, E.; Shi, L.; Wang, M.; Wang, B. M.; Zhang, S. Y. *Org. Lett.* **2004**, 6, 4691–4694. (m) Liang, P. H.; Liu, J. P.; Hsin, L. W.; Cheng, C. Y. *Tetrahedron* **2004**, 60, 11655–11660.

(10) Racemic syntheses (narrowed): (a) Holton, R. A.; Sibi, M. P.; Murphy, W. S. *J. Am. Chem. Soc.* **1988**, 110, 314–316. (Galanthamine): (b) Guillou, C.; Beunard, J. L.; Gras, E.; Thal, C. *Angew. Chem., Int. Ed.* **2001**, 40, 4745–4746. (c) Hu, X.-D.; Tu, Y.-Q.; Zhang, E.; Gao, S.; Wang, S.; Wang, A.; Fan, C.-A.; Wang, M. *Org. Lett.* **2006**, 8, 1823–1825.

(11) Synthetic approaches/formal syntheses: (a) Parsons, P. J.; Charles, M. D.; Harvey, D. M.; Sumoreeah, L. R.; Shell, A.; Spoors, G.; Gill, A. L.; Smith, S. *Tetrahedron Lett.* **2001**, 42, 2209–2211. (b) Pilger, C.; Westermann, B.; Florke, U.; Fels, G. *Synlett* **2000**, 1163–1165.

(12) Trost, B. M.; Tang, W. P.; Toste, F. D. *J. Am. Chem. Soc.* **2005**, 127, 14785–14803.

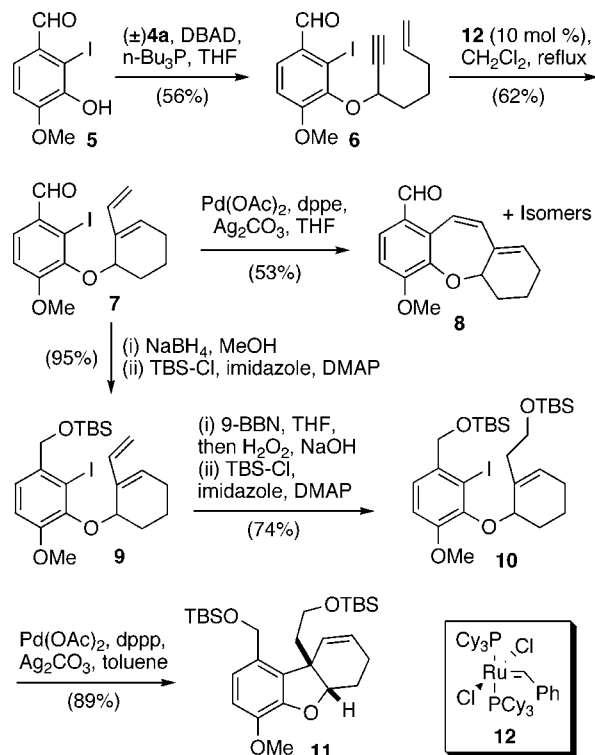
(13) For some examples of galanthamine analogues, see: (a) Bores, G. M.; Kosley, R. W. *J. Drugs Future* **1996**, 21, 621–635. (b) Poschalko, A.; Welzig, S.; Treu, M.; Nerdinger, S.; Mereiter, K.; Jordis, U. *Tetrahedron* **2002**, 58, 1513–1518. (c) Pelish, H. E.; Westwood, N. J.; Feng, Y.; Kirchhausen, T.; Shair, M. D. *J. Am. Chem. Soc.* **2001**, 123, 6740–6741. (d) Herlem, D.; Martin, M. T.; Thal, C.; Guillou, C. *Bioorg. Med. Chem. Lett.* **2003**, 13, 2389–2391. (e) Treu, M.; Jordis, U.; Mereiter, K. *Heterocycles* **2001**, 55, 1727–1735.

(14) Enyne metathesis: (a) Diver, S. T.; Giessert, A. *J. Chem. Rev.* **2004**, 104, 1317–1382. (b) Poulsen, C. S.; Madsen, R. *Synthesis* **2003**, 1–18.

(15) Kiankarimi, M.; Lowe, R.; McCarthy, J. R.; Whitten, J. P. *Tetrahedron Lett.* **1999**, 40, 4497–4500.

Our initial investigations, using a racemic propargylic alcohol (±)-**4a**¹⁷ and phenol **5**,¹⁸ established the enyne RCM reaction as a highly chemoselective means to create the vinylcyclohexene **7** (Scheme 2). However, it transpired that

Scheme 2. Synthesis of a Racemic ABD Ring System



Pd-catalyzed arylation of the diene **7** did not provide the desired product and instead favored C–C bond formation at the less sterically encumbered end of the 1,3-diene system, affording diene **8**.¹⁹ Presumably, the intermediacy of a π -allyl palladium species also favors the observed regioselectivity. The required C–C bond formation was realized by selective hydration of the less-substituted olefin prior to the Heck reaction, securing the tricyclic ABD system **11** in high overall yield.^{11,12,20,21}

Although the ultimately successful route to the tricycle **11** came at the expense of additional functional group interconversions and protecting group manipulations, it was easy to envisage its adaptation to provide a streamlined asymmetric synthesis of galanthamine. Accordingly, the

(16) For the synthesis of protected **4b**, see: Smith, A. B.; Ott, G. R. *J. Am. Chem. Soc.* **1998**, 120, 3935–3948.

(17) Padwa, A.; Lipka, H.; Watterson, S. H.; Murphree, S. S. *J. Org. Chem.* **2003**, 68, 6238–6250.

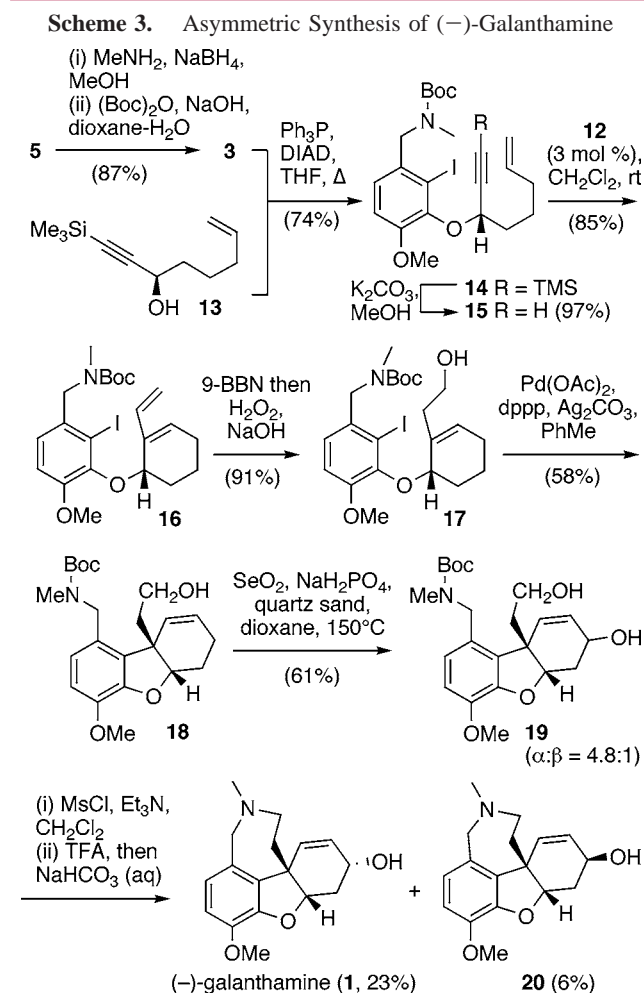
(18) Tantishaiyakul, V.; Hamada, A.; Miller, D. D.; Romstedt, K. J.; Shams, G.; Shin, Y.; Fraundorfer, P. F.; Doyle, K.; Feller, D. R. *J. Med. Chem.* **1992**, 35, 466–479.

(19) For a discussion of intermolecular palladium-catalyzed diene arylation, see: Heck, R. F. *Palladium Reagents in Organic Syntheses*; Academic Press: New York, 1985; pp 223–227.

(20) (a) Gibson, S. E.; Middleton, R. J. *Contemp. Org. Synth.* **1996**, 3, 447–471. (b) Overman, L. E. *Pure Appl. Chem.* **1994**, 66, 1423–1430.

(21) For a related application of the Heck reaction, see: Uchida, K.; Yokoshima, S.; Kan, T.; Fukuyama, T. *Org. Lett.* **2006**, 8, 5311–5313.

protected aminomethyl group was introduced into the aryl fragment **3** (Scheme 3). Mitsunobu coupling of **3** with the



enantiomerically enriched propargylic alcohol **13** delivered the aryl ether **14**. The enyne **15** (92% ee, HPLC), obtained after TMS deprotection of **14**, underwent an efficient RCM reaction in the presence of 3 mol % of the first-generation Grubbs' catalyst (**12**) at ambient temperature to give diene **16**. Hydroboration and oxidation of **16** was unaffected by the presence of the carbamate group, allowing access to the homoallylic alcohol **17** in excellent yield (91%).

The remaining steps to complete the total synthesis of (–)-galanthamine were carried out without protection of the primary hydroxyl group. An intramolecular Heck reaction of **17** led to successful formation of the central heterocyclic five-membered D ring in an acceptable yield. Trost's procedure was then applied to install the allylic hydroxyl group into **19**,¹² leading to an excess of the desired α -diastereoisomer (dr = 4.8:1 estimated by ^1H NMR). Separation of the diastereoisomers was not possible at this stage, so the mixture of epimers was taken through the remaining two steps. Activation of the 1° hydroxyl group was achieved through mesylation. Although some bismesylation was also observed under the conditions employed, no attempt was made to optimize this reaction. Finally, the azepine B ring formation was achieved in a one-pot process by sequential treatment with TFA and neutralization with NaHCO_3 (aq). Thus (–)-galanthamine and its epimer **19** were obtained after separation by column chromatography.²²

In conclusion, an enantioselective synthesis of (–)-galanthamine has been described from commercially available materials in 11 linear steps (11 steps from isovanillin or 5-hexenoic acid, 14 steps in total). Control of absolute stereochemistry was achieved through an asymmetric reduction of a propargylic ketone. An efficient enyne metathesis reaction was used to close the B ring while generating the requisite functionality later used in the formation of the D and C rings.

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Supporting Information Available: Procedures and spectroscopic and analytical data for compounds **1**, **3**, **4a**, **6–11**, and **13–19**. Copies of ^1H and ^{13}C NMR spectra for compounds **1**, **3**, **6–11**, and **13–19**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(22) Spectroscopic and physical data for synthetic (–)-galanthamine were consistent with those reported in the literature: ref 12 for ^1H and ^{13}C NMR. Melting point and $[\alpha]_D$: ref 8 and: Kobayashi, S.; Yuasa, K.; Sato, K.; Imakura, Y.; Shingu, T. *Heterocycles* **1982**, *19*, 1219–1222.